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~~Progress of Chemical Reactions~~

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a state of balance in which the rates of the forward and reverse reactions are equal; no net change in the amount of reactants and products occurs in the chemical system (18.2)

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The rate of a chemical reaction is expressed as the amount of reactant changing per unit time.

Work Step by Step The rate of a chemical reaction is expressed as the amount of reactant changing

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CHAPTER 18. Reaction Rates and Equilibrium. 18.1 Rates of Reaction. •The time needed to complete a chemical reaction can vary tremendously. •Rates measure changes that occur within intervals of time. •In chemistry, rates or speeds are

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expressed in chemical terms.

Atoms, ions, and molecules must collide in order to react. Atoms, ions, and molecules can form a chemical bond when they collide, as long as the particles have enough kinetic energy and have the proper orientation.

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atoms, ions, and molecules can
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collide, provided that the particles
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activation energy.

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