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Uranium 238 has a half-life of  $4.51 \times 10^9$  years, whereas  $^{235}\text{U}$  has a half-life of  $7.1 \times 10^8$  years. The natural abundance of  $^{238}\text{U}$  in a sample of uranium is

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99.2739%, and that of  $^{235}\text{U}$  is 0.7205%.  
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The answer is solved by creating the fraction.  $n \cdot 2^{-1}$ . Where  $n$  = the number of half lives. If each half life is 5 seconds, then in one minute. (60 seconds) there are 12 half lives.

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After the third half-life, how much of the sample is left? Preview this quiz on Quizizz. After the third half-life, how much of the sample is left? Half-Life DRAFT. 9th - 12th grade. ... 15 Questions Show answers. Question 1 . SURVEY . 120 seconds . Q. After the third half-life, how much of the sample is left? answer

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choices .  $1/2$ .  $1/3$ .  $1/16$ . 1 ...

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If the half-life is 23.9 hours, how much will be present at the end of one day? Two days? Seven days? Solution:  $24.0 \text{ hr} / 23.9 \text{ hr/half-life} = 1.0042$  half-lives One day = one half-life;  $(1/2)^{1.0042} = 0.4985465$  remaining = 4.98 g Two days = two half-lives;  $(1/2)^{2.0084} = 0.2485486$  remaining

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= 2.48 g

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ChemTeam: Half-Life Problems #1 - 10

Half-life, the amount of time it takes for one half of a radioactive substance to decay, varies for different isotopes.

Develop models to illustrate the changes in

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the composition of the nucleus of the atom and the energy released during the processes of fission, fusion, and radioactive decay.

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What is the half-life for a particular reaction if the rate law is  $\text{rate} = (1.301 \text{ M} \cdot \text{min}^{-1})[\text{A}]^0$  and the initial concentration of A is 0.250 M? Show transcribed image text Expert Answer

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Solved: What Is The Half-life For A

*Page 24/36*



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Particular Reaction If ...

If the half-life of nitrogen-13 is 10 minutes.

A) What fraction of nitrogen-13 will be left after 20 minutes? (show or explain your answer) B) What fraction of nitrogen-13 will be left after 50 minutes? (show or explain your answer) Please help me figure out the answers to these 2 questions.

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Thank you for your time.

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Solved: If The Half-life Of Nitrogen-13 Is 10 Minutes. A ...

The half-life of Zn-71 is 2.4 minutes. If one had 100.0 g at the beginning, how many grams would be left after 7.2

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minutes has elapsed? answer choices

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The half-life of a radioactive isotope is known to be exactly 1 h. a) What fraction of a sample would be left after exactly 3 h?

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% Remaining = % b) What fraction of a sample would be left after exactly 1 day?

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Solved: The Half-life Of A Radioactive Isotope Is Known To ...

Cobalt-60 is radioactive and has a half life of 5.26 years. Calculate the activity of a

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8.1 mg sample of cobalt-60. Give your answer in becquerels and in curies. Be sure your answer has the correct number of significant digits. Bq    \$

---

Solved: Cobalt-60 Is Radioactive And Has A Half Life Of 5 ...

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4) What is the half-life of isotope Z if a rock contains a parent isotope: daughter product ratio of 1:7 and the rock is 9 million years old? How do you know this? Show your work for full credit.

---

4) What Is The Half-life Of Isotope Z If A

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Rock Co ...

What is the Half-life of a Drug? The half-life of a drug is an estimate of the period of time that it takes for the concentration or amount in the body of that drug to be reduced by exactly one half (50%). The symbol for half-life is  $T_{1/2}$ . For example, if 100mg of a drug with a half-life of 60

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minutes is taken, the following is estimated:

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Drug Half-life Explained: Calculator, Variables & Examples

How many "half-lives" are there in 2 full days. 2 days is 48 hours.  $48 / 8 = 6$  So the



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amount of the sample is cut in half 6 times.  
 $1.6 \times (1/2)^6 = 1.6 / 64 = 0.025$  kg or 25 g. You can do it like...

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Isotope X has a half-life of 8 hours. How ... - Yahoo Answers

Answered: A particular isotope has a

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half-life... | bartleby A particular isotope has a half-life of 77 77 days. If you start with 1 1 kilogram of this isotope, how much will remain after 150 150 days? after 300 300 days? The remaining amount after 150 150 days will be nothing kilogram (s).

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Answered: A particular isotope has a half-life... | bartleby

Please answer the following question. Half life of the radium isotope is approximately 1,599 years ---- IF the amount left after 1k years is 1.7 grams, what is the amt after 2k years? --Please leave the answer in nearest

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tenth thousandths

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