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Like the decelerating car, the average rate of a chemical reaction will fall somewhere between its initial and final rates. The instantaneous rate of a reaction may be determined one of two ways. If

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experimental conditions permit the measurement of concentration changes over very short time intervals, then average rates computed as described earlier provide reasonably good approximations of ...

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A reversible chemical reaction is in chemical equilibrium when the rate of its forward reaction equals the rate of its reverse reaction and the concentrations of its products and reactants remain

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unchanged. The chemical equation for the reaction at equilibrium is written using double arrows to indicate the overall reversibility of the reaction.

~~CHAPTER 18 Chemical Equilibrium~~

Chemistry Paper Two - Rates of Reaction.

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change in temperature or the giving off of light energy □ a color change in the reaction system ...

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is formed. There are different ways to determine the rate of a reaction. The method chosen usually depends on ...

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Reactions Section: Rates of Change 1. State Le Châtelier's principle. 2. Describe two ways you could make table salt dissolve faster in water. 3. Predict the shift of equilibrium for each of the following conditions in the following reaction involving gaseous reactants and products:.

a.  $\text{NH}_3$  is added to the ...

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